Organic Chemistry II, Fall 2019, Quiz 1a

Name:

Signature:

The following quiz will begin 5 minutes into recitation and will last for 30 minutes. If you use back of paper, please make note of it for credit. No leaving for any reason once the quiz begins. Good luck!

1. Name the reactants, reagents and/or products of the following reactions (10 points, 2.5 points each)

a)
$$Br_2$$

c)
$$H_2SO_4$$

a)
$$H_2SO_4$$
 OF

Signature:

The following quiz will begin 5 minutes into recitation and will last for 30 minutes. If you use back of paper, please make note of it for credit. No leaving for any reason once the quiz begins. Good luck!

1. Name the reactants, reagents and/or products of the following reactions (10 points, 2.5 points each)

Organic Chemistry II, Fall 2019, Quiz 1c

Name:

Signature:

The following quiz will begin 5 minutes into recitation and will last for 30 minutes. If you use back of paper, please make note of it for credit. No leaving for any reason once the quiz begins. Good luck!

1. Name the reactants, reagents and/or products of the following reactions (10 points, 2.5 points each)

d)
$$CrO_3$$
, H_2SO_4

a)
$$H_2SO_4$$
 O H_2O

Organic Chemistry II, Fall 2019, Quiz 1d

Name:

Signature:

The following quiz will begin 5 minutes into recitation and will last for 30 minutes. If you use back of paper, please make note of it for credit. No leaving for any reason once the quiz begins. Good luck!

1. Name the reactants, reagents and/or products of the following reactions (10 points, 2.5 points each)

c)
$$CrO_3/H_2SO_4$$

a)
$$H_3PO_4$$

Organic Chemistry II, Fall 2019, Quiz 1a

Name:

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Signature:

The following quiz will begin 5 minutes into recitation and will last for 30 minutes. If you use back of paper, please make note of it for credit. No leaving for any reason once the quiz begins. Good luck!

1. Name the reactants, reagents and/or products of the following reactions (10 points, 2.5 points each)

a)
$$Br_2$$
 Br_3

c) OH
$$H_2SO_4$$

a)
$$H_2SO_4$$
 H_2O
 H

Organic Chemistry II, Fall 2019, Quiz 1

Name:

Key

Signature:

The following quiz will begin 5 minutes into recitation and will last for 30 minutes. If you use back of paper, please make note of it for credit. No leaving for any reason once the quiz begins. Good luck!

1. Name the reactants, reagents and/or products of the following reactions (10 points, 2.5 points each)

a)
$$\begin{array}{c}
Br_{2} \\
MeOH
\end{array}$$

$$\begin{array}{c}
Br_{2} \\
MeOH
\end{array}$$

$$\begin{array}{c}
Br_{2} \\
Br
\end{array}$$

$$\begin{array}{c}
Arrow \\
Br$$

$$\begin{array}{c}
Arrow \\
Br
\end{array}$$

$$\begin{array}{c}
Arrow \\
Br
\end{array}$$

$$\begin{array}{c}
Arrow \\
Br$$

$$\begin{array}{c}
Arrow \\
Br
\end{array}$$

$$\begin{array}{c}
Arrow \\
Br
\end{array}$$

$$\begin{array}{c}
Arrow \\
Br$$

$$\begin{array}{c}
Arrow \\
Br
\end{array}$$

$$\begin{array}{c}
Arrow \\
Br$$

$$\begin{array}{c}
Arrow \\
Br$$

$$Arrow \\
Br$$

$$\begin{array}{c}
Arrow \\
Br$$

$$Arrow \\
Br$$

$$\begin{array}{c}
Arrow \\
Br$$

$$Arrow \\
Br$$

$$Arrow$$

Organic Chemistry II, Fall 2019, Quiz 1c

Name:

Signature:

The following quiz will begin 5 minutes into recitation and will last for 30 minutes. If you use back of paper, please make note of it for credit. No leaving for any reason once the quiz begins. Good luck!

1. Name the reactants, reagents and/or products of the following reactions (10 points, 2.5 points each)

a)
$$BH_3$$
; NaOOH $(-1)^{-1}$ OH

b)
$$OH \longrightarrow OH$$

d)
$$OH$$

a)
$$H_{2}SO_{4}$$

$$H_{2}SO_{4}$$

$$H_{2}O$$

$$HO_{5}H$$

$$O_{5}H$$

$$O_{7}H$$

$$O_{7}$$

Organic Chemistry II, Fall 2019, Quiz 1d

Name:

Signature:

The following quiz will begin 5 minutes into recitation and will last for 30 minutes. If you use back of paper, please make note of it for credit. No leaving for any reason once the quiz begins. Good luck!

1. Name the reactants, reagents and/or products of the following reactions (10 points, 2.5 points each)

c)
$$CrO_3/H_2SO_4$$
 OM

a) Ph
$$=$$
 $\frac{i. \text{ NaNH}_2}{ii. \text{ MeBr}}$ Ph $=$ Me $\frac{i. \text{ NaNH}_2}{ii. \text{ MeBr}}$ $\frac{i. \text{ NaNH}_2}{ii. \text{ NaNH}_2}$ $\frac{i$