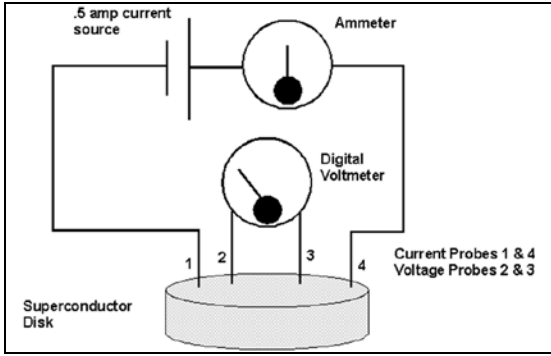
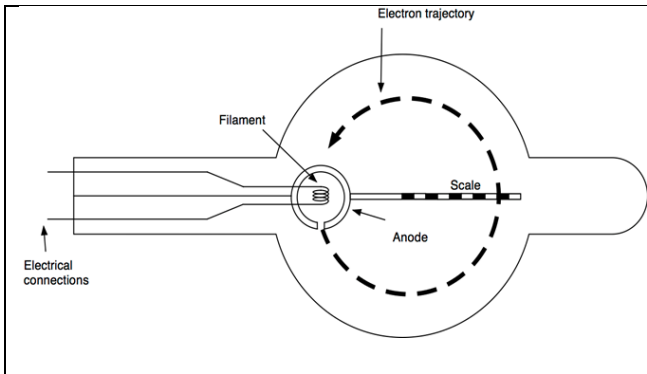


1. (Superconductivity, 25%)



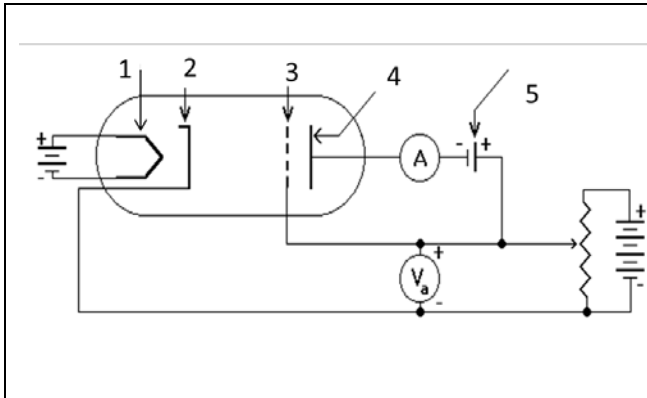
- a. As shown in left, a four probe arrangement was used in our superconductivity experiment. Indicate the current-source, voltage-meter, and ampere-meter in the diagram.
- b. If 0.5 A current is injected into the superconductor under superconducting state, how many cooper pairs are been transported per second from the terminal "1" to "4"?
- Ans. 1.56 x 10¹⁸

2. (Charge-mass ratio of an electron 25%)

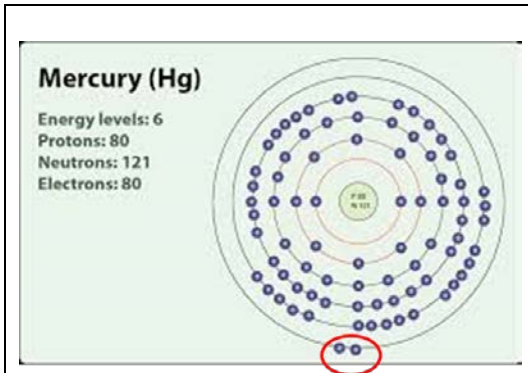


- a. Indicate the direction of magnetic field either into or out from the paper based on the electron trajectory as shown in the left diagram.
- Ans. **B** out from the paper
- b. Estimate value for the error δR due to the fuzziness of the measured radius, $\delta r = 0.001$ m, with the assumptions of $V = 100$ volt, $r = 0.01$ m, and $B = 0.5$ T.
- Ans. - 3.2 x 10⁶ C/kg

3. (Frank-Hertz, Hg excited by energetic e-, 25%)

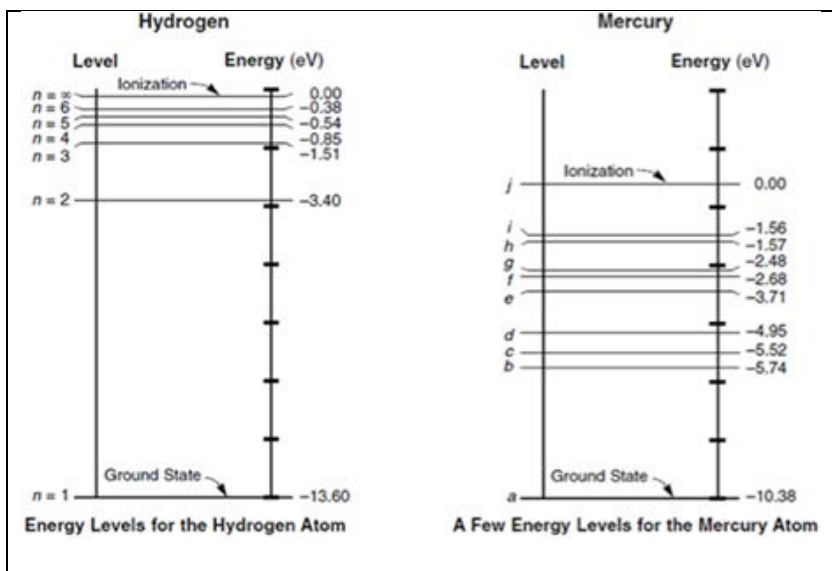


- a. Identify the parts of 2 to 5 in the schematic diagram shown in left, used for our Frank-Hertz experiment:
1. heating filament
 2. Cathode
 3. Anode (grid)
 4. Collector
 5. Retarding bias (voltage)
- b. Can $|V_{4-5}| > |V_{3-2}|$ in your experiment? Ans. Yes No X

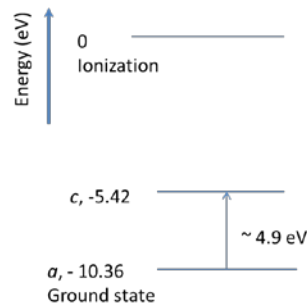


- c. What is the value of principle quantum number, n?
- Ans. 6
- d. Circle the electrons involved in the inelastic scattering during your experiments in the schematic diagram shown in left.

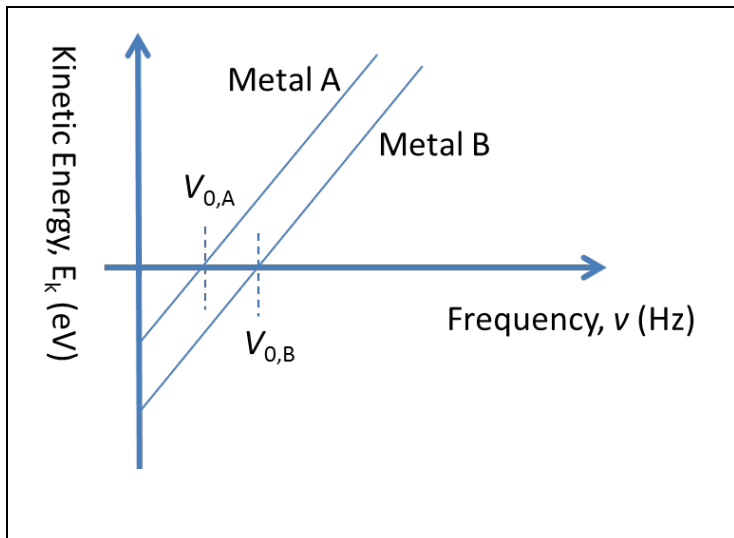
| | |
|--|--|
| | |
|--|--|



e. Draw (below) an energy transition diagram for the electrons involved when Hg atoms are excited by energetic electrons based on the energy-diagram shown in left.



4. (Photoelectric effect, 25%)



Given:
 $V_{0,A} = 5.3 \times 10^{14} \text{ Hz}$
 $V_{0,B} = 5.6 \times 10^{14} \text{ Hz}$

Referring to the diagram shown in left as the measured photoelectric data (maximum kinetic energy vs. frequency) from the two metals (A and B), find:

- the work-function for the metal A in eV,
Ans. 2.2 eV
- the work-function for the metal B in eV,
Ans. 2.3 eV

Reference equations and constants:

| | |
|-------------------------|---|
| Charge-mass ratio, R | $R = e/m = 2V/(r^2B^2)$; The recommended value of R for an <i>electron</i> is $e/m = -1.758820024 (11) \times 10^{11} \text{ C/kg}$. |
| Photoelectric effect | Planck constant, $h = 6.626 \times 10^{-34} \text{ J.s}$. $1 \text{ J} = 6.242 \times 10^{18} \text{ eV}$ |
| Elementary charge value | $e^- = -1.6 \times 10^{-19} \text{ C (coulombs)}$ |